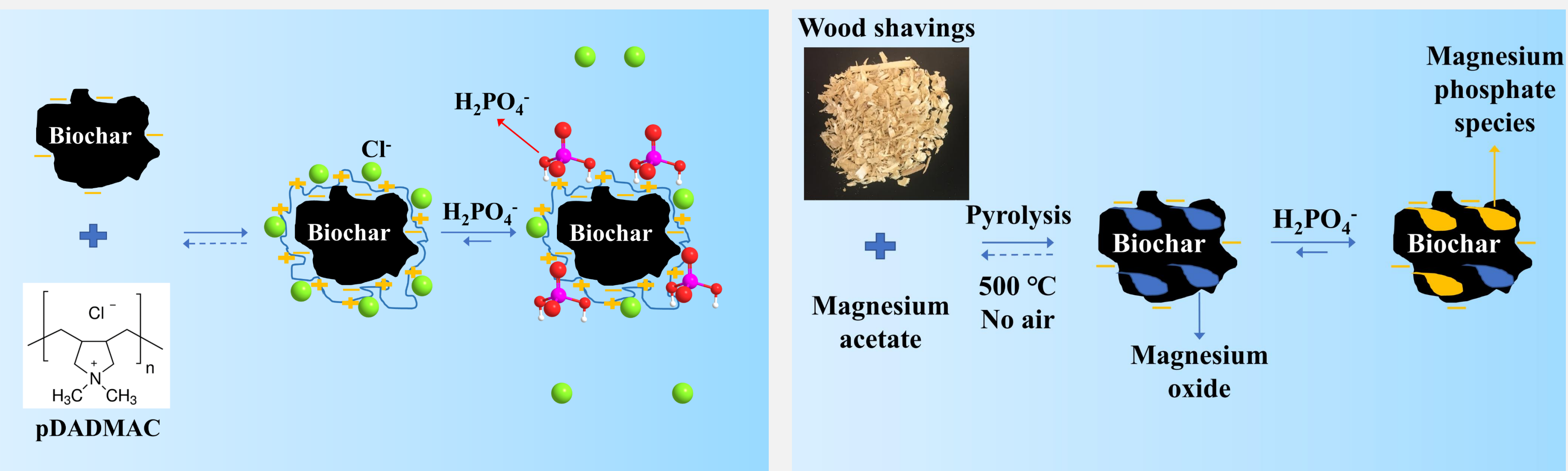


Introduction

- Manures contain excess nutrients in highly leachable forms.
- The objective was to design modified biochars that bind excess nutrients, rendering them less leachable but still bio-available to plants after the composted manure is applied to fields.
- One modification is coating with poly dimethyldiallyl ammonium chloride (pDADMAC), a cationic polymer that reverses the biochar surface charge and attracts phosphate anions.
- Another modification is coating with nano-films of magnesium oxide, which can strongly bind phosphate.
- We examined binding and release of ortho phosphate (salts of H_3PO_4) and the effects of natural dissolved substances in manure waters.



Methodology

- Biochar is made by heating vegetation wastes at high temperature without air (in this case, wood shavings at 500 °C).
- The polymer-coated biochar was made by mixing the biochar with a water solution of the polymer (pDADMAC).
- Magnesium-coated biochar was made by first soaking the wood shavings in a water solution of magnesium acetate before charring.
- Binding of phosphate was measured using a 3-day equilibration time.
- Release of phosphate was studied by removing a portion of the liquid and adding pure water, then allowing re-equilibration for 3 days. This was repeated 1-3 times.

Results and Conclusions

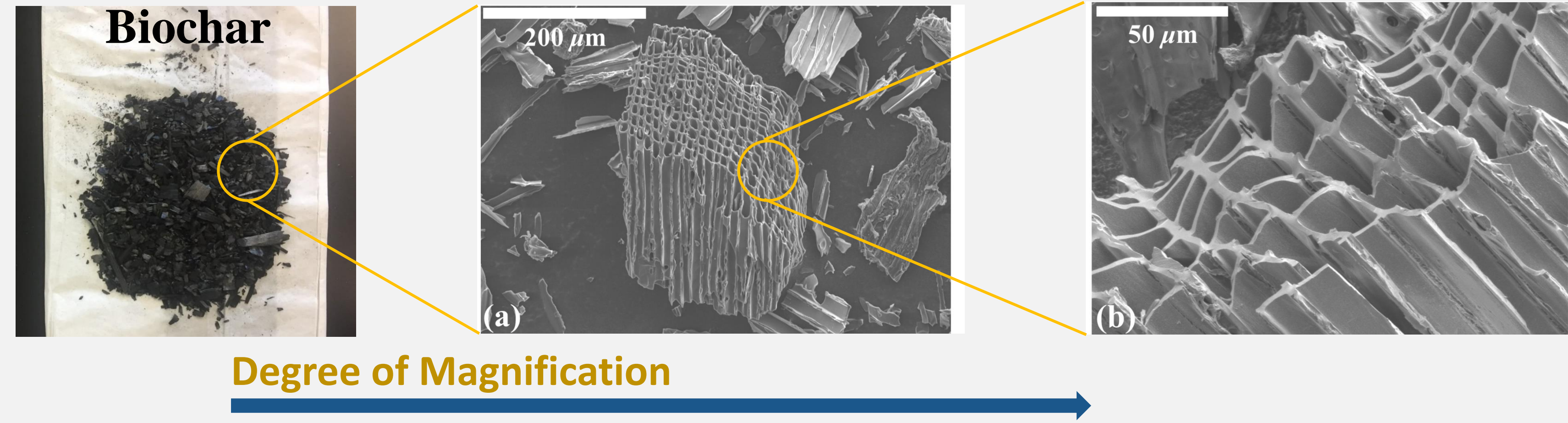


Figure 1. A picture and scanning electron microscopic images of pDADMAC-coated biochar showing its highly porous nature.

Figure 2. Binding of phosphate increases dramatically with increasing coating density of pDADMAC or magnesium (Mg) oxide.

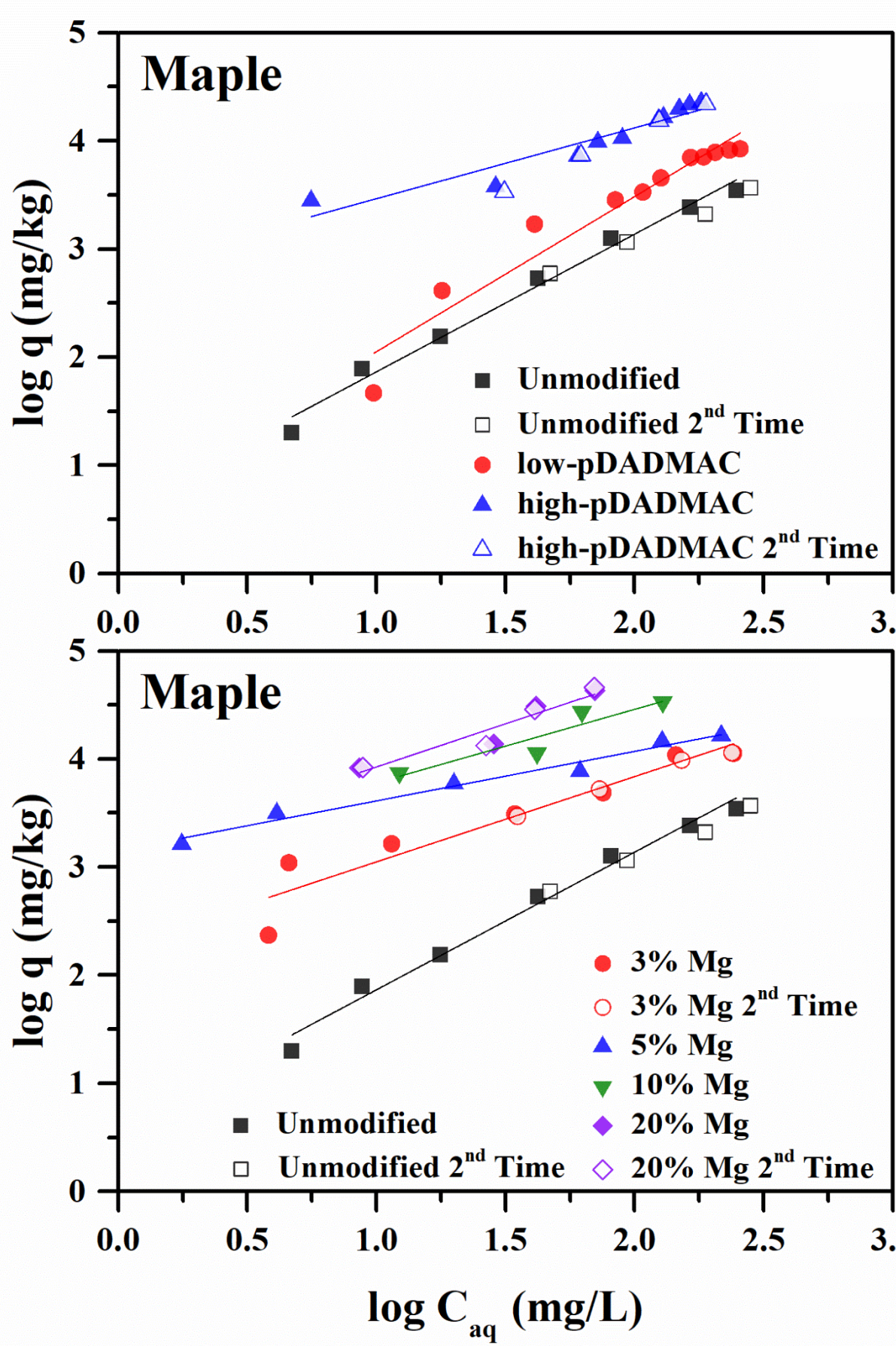


Figure 3. Phosphate is less reversibly-bound—therefore, less bio-available—by magnesium (Mg)-coated than pDADMAC-coated biochars.

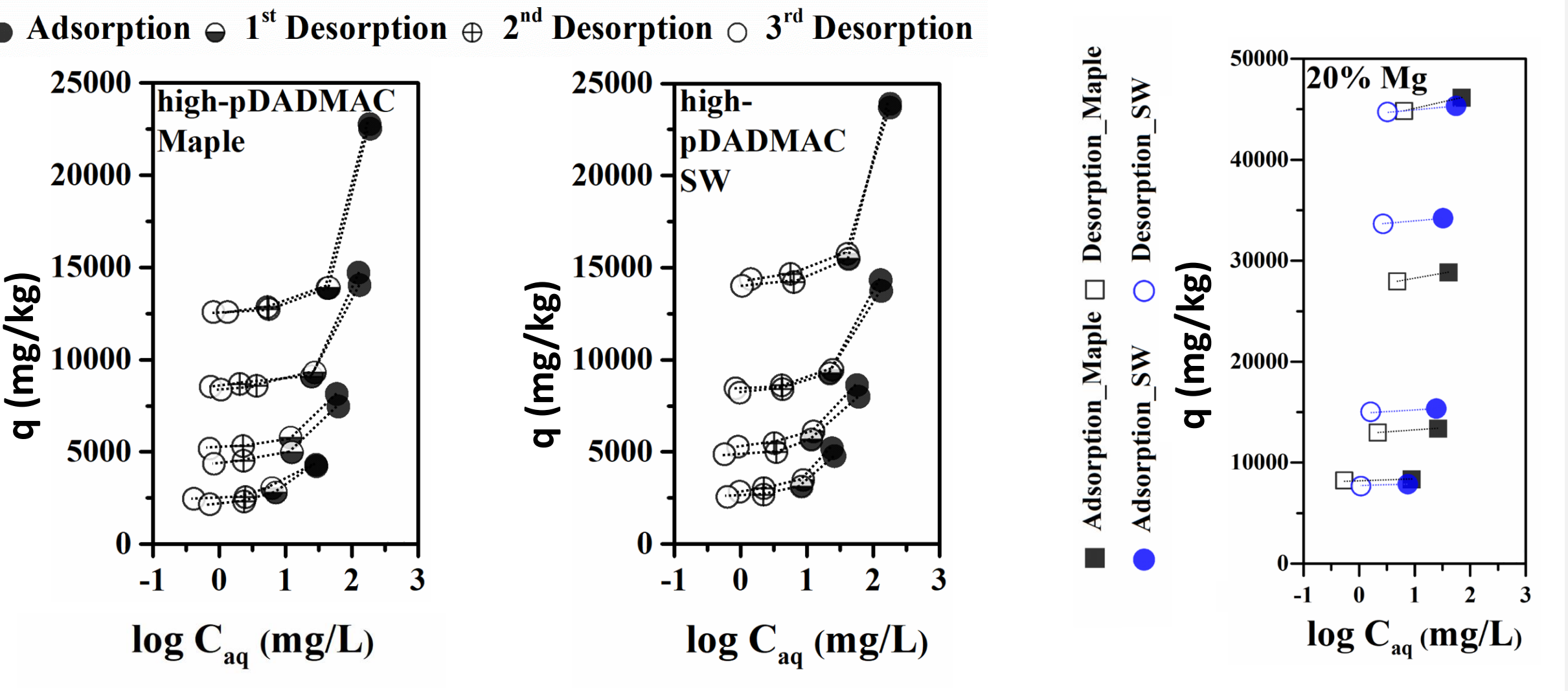
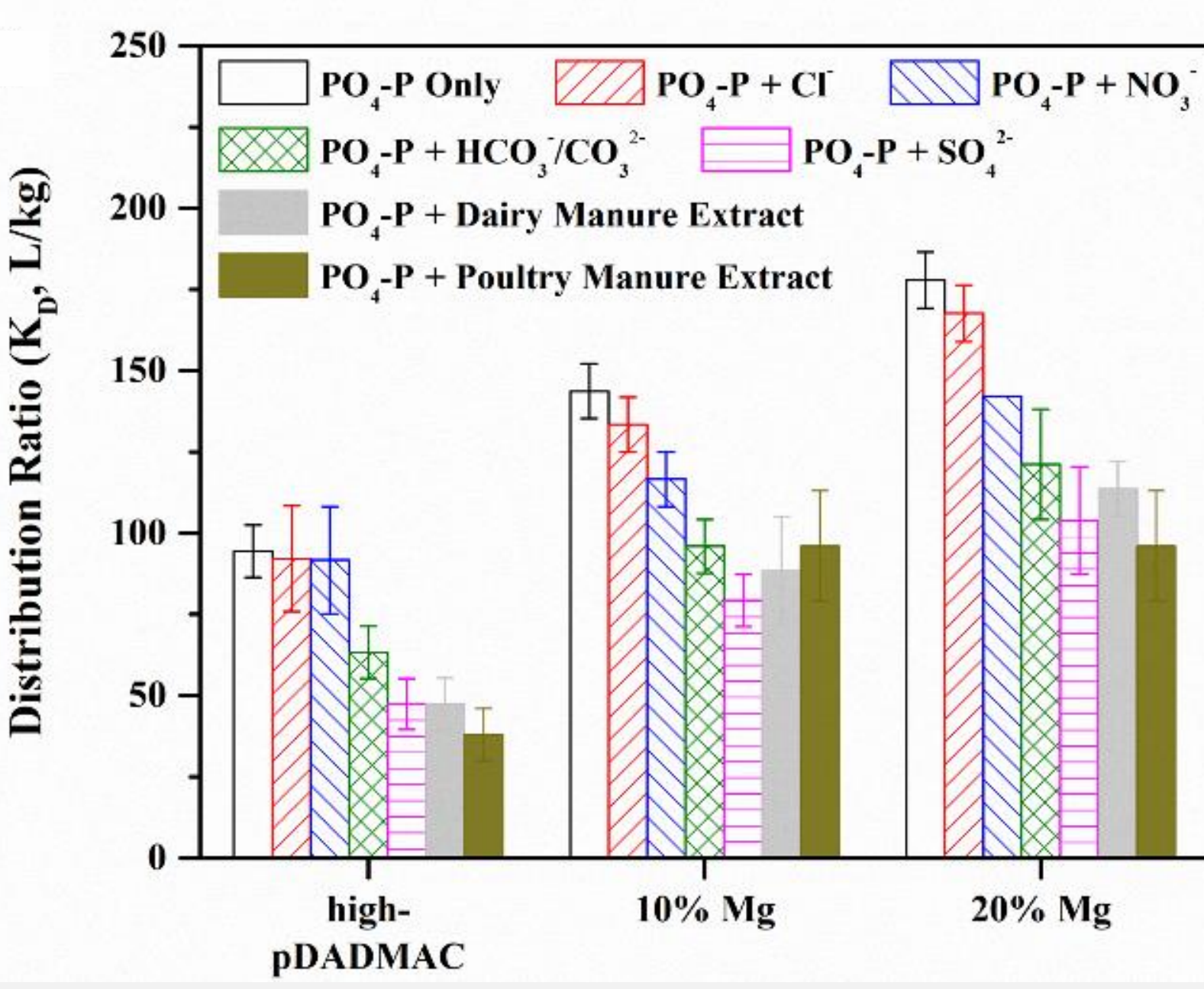


Figure 4. Binding of phosphate is unaffected by nitrate or chloride, but moderately inhibited by carbonate, sulfate, and dairy manure extract.



(plotted on logarithmic scale)

Future Work: 1) Modified biochars are being tested in potting trials.
2) Removal & recycling strategies for ammonium and nitrate will be studied.

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